

Preface

Fundamental Physical Constants: 1998

Peter J. Mohr and Barry N. Taylor

National Institute of Standards and Technology, Gaithersburg, MD 20899-8401

This table gives the 1998 self-consistent set of values of the basic constants and conversion factors of physics and chemistry recommended by the Committee on Data for Science and Technology (CODATA) for international use. Further, it describes in detail the adjustment of the values of the subset of constants on which the complete 1998 set of recommended values is based. The 1998 set replaces its immediate predecessor recommended by CODATA in 1986. The new adjustment, which takes into account all of the data available through 31 December 1998, is a significant advance over its 1986 counterpart. The 1998 adjustment was carried out by P. J. Mohr and B. N. Taylor of the National Institute of Standards and Technology (NIST) under the auspices of the CODATA Task Group on Fundamental Constants. The standard uncertainties (i.e., estimated standard deviations) of the new recommended values are in most cases about 1/5 to 1/12 and in some cases 1/160 times the standard uncertainties of the corresponding 1986 values. Moreover, in almost all cases the absolute values of the differences between the 1998 values and the corresponding 1986 values are less than twice the standard uncertainties of the 1986 values.

The Task Group was established in 1969 with the aim of periodically providing the scientific and technological communities with a self-consistent set of internationally recommended values of the fundamental physical constants based on all applicable information available at a given point in time. The first set was published in 1973 and was followed by a revised set first published in 1986; the current 1998 set first appeared in 1999. In the future, the CODATA Task Group plans to take advantage of the high level of automation developed for the current set in order to issue a new set of recommended values at least every four years.

Fundamental Physical Constants — Atomic and Nuclear Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
General				
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	α	$7.297\ 352\ 533(27) \times 10^{-3}$		3.7×10^{-9}
inverse fine-structure constant	α^{-1}	$137.035\ 999\ 76(50)$		3.7×10^{-9}
Rydberg constant $\alpha^2 m_e c / 2h$	R_∞	$10\ 973\ 731.568\ 549(83)$	m^{-1}	7.6×10^{-12}
	$R_\infty c$	$3.289\ 841\ 960\ 368(25) \times 10^{15}$	Hz	7.6×10^{-12}
	$R_\infty hc$	$2.179\ 871\ 90(17) \times 10^{-18}$	J	7.8×10^{-8}
$R_\infty hc$ in eV		$13.605\ 691\ 72(53)$	eV	3.9×10^{-8}
Bohr radius $\alpha/4\pi R_\infty = 4\pi\epsilon_0\hbar^2/m_e e^2$	a_0	$0.529\ 177\ 2083(19) \times 10^{-10}$	m	3.7×10^{-9}
Hartree energy $e^2/4\pi\epsilon_0 a_0 = 2R_\infty hc$ $= \alpha^2 m_e c^2$ in eV	E_h	$4.359\ 743\ 81(34) \times 10^{-18}$ $27.211\ 3834(11)$	J eV	7.8×10^{-8} 3.9×10^{-8}
quantum of circulation	$h/2m_e$	$3.636\ 947\ 516(27) \times 10^{-4}$	$\text{m}^2 \text{s}^{-1}$	7.3×10^{-9}
	h/m_e	$7.273\ 895\ 032(53) \times 10^{-4}$	$\text{m}^2 \text{s}^{-1}$	7.3×10^{-9}
Electroweak				
Fermi coupling constant ^a	$G_F/(\hbar c)^3$	$1.166\ 39(1) \times 10^{-5}$	GeV^{-2}	8.6×10^{-6}
weak mixing angle ^b θ_W (on-shell scheme) $\sin^2 \theta_W = s_W^2 \equiv 1 - (m_W/m_Z)^2$	$\sin^2 \theta_W$	0.2224(19)		8.7×10^{-3}
Electron, e^-				
electron mass in u, $m_e = A_r(e)$ u (electron relative atomic mass times u)	m_e	$9.109\ 381\ 88(72) \times 10^{-31}$	kg	7.9×10^{-8}
energy equivalent in MeV	$m_e c^2$	$5.485\ 799\ 110(12) \times 10^{-4}$ $8.187\ 104\ 14(64) \times 10^{-14}$ $0.510\ 998\ 902(21)$	u J MeV	2.1×10^{-9} 7.9×10^{-8} 4.0×10^{-8}
electron-muon mass ratio	m_e/m_μ	$4.836\ 332\ 10(15) \times 10^{-3}$		3.0×10^{-8}
electron-tau mass ratio	m_e/m_τ	$2.875\ 55(47) \times 10^{-4}$		1.6×10^{-4}
electron-proton mass ratio	m_e/m_p	$5.446\ 170\ 232(12) \times 10^{-4}$		2.1×10^{-9}
electron-neutron mass ratio	m_e/m_n	$5.438\ 673\ 462(12) \times 10^{-4}$		2.2×10^{-9}
electron-deuteron mass ratio	m_e/m_d	$2.724\ 437\ 1170(58) \times 10^{-4}$		2.1×10^{-9}
electron to alpha particle mass ratio	m_e/m_α	$1.370\ 933\ 5611(29) \times 10^{-4}$		2.1×10^{-9}
electron charge to mass quotient	$-e/m_e$	$-1.758\ 820\ 174(71) \times 10^{11}$	C kg^{-1}	4.0×10^{-8}
electron molar mass $N_A m_e$	$M(e), M_e$	$5.485\ 799\ 110(12) \times 10^{-7}$	kg mol^{-1}	2.1×10^{-9}
Compton wavelength $h/m_e c$ $\lambda_C/2\pi = \alpha a_0 = \alpha^2/4\pi R_\infty$	λ_C	$2.426\ 310\ 215(18) \times 10^{-12}$ $386.159\ 2642(28) \times 10^{-15}$	m	7.3×10^{-9} 7.3×10^{-9}
classical electron radius $\alpha^2 a_0$	r_e	$2.817\ 940\ 285(31) \times 10^{-15}$	m	1.1×10^{-8}
Thomson cross section $(8\pi/3)r_e^2$	σ_e	$0.665\ 245\ 854(15) \times 10^{-28}$	m^2	2.2×10^{-8}
electron magnetic moment to Bohr magneton ratio	μ_e	$-928.476\ 362(37) \times 10^{-26}$	J T^{-1}	4.0×10^{-8}
to nuclear magneton ratio	μ_e/μ_B	$-1.001\ 159\ 652\ 1869(41)$		4.1×10^{-12}
electron magnetic moment anomaly $ \mu_e /\mu_B - 1$	μ_e/μ_N	$-1\ 838.281\ 9660(39)$		2.1×10^{-9}
	a_e	$1.159\ 652\ 1869(41) \times 10^{-3}$		3.5×10^{-9}

Fundamental Physical Constants — Atomic and Nuclear Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
electron g-factor $-2(1 + a_e)$	g_e	$-2.002\,319\,304\,3737(82)$		4.1×10^{-12}
electron-muon				
magnetic moment ratio	μ_e/μ_μ	206.766 9720(63)		3.0×10^{-8}
electron-proton				
magnetic moment ratio	μ_e/μ_p	$-658.210\,6875(66)$		1.0×10^{-8}
electron to shielded proton				
magnetic moment ratio (H ₂ O, sphere, 25 °C)	μ_e/μ'_p	$-658.227\,5954(71)$		1.1×10^{-8}
electron-neutron				
magnetic moment ratio	μ_e/μ_n	960.920 50(23)		2.4×10^{-7}
electron-deuteron				
magnetic moment ratio	μ_e/μ_d	$-2\,143.923\,498(23)$		1.1×10^{-8}
electron to shielded helion ^c				
magnetic moment ratio (gas, sphere, 25 °C)	μ_e/μ'_h	864.058 255(10)		1.2×10^{-8}
electron gyromagnetic ratio $2 \mu_e /\hbar$	γ_e	$1.760\,859\,794(71) \times 10^{11}$	$s^{-1} T^{-1}$	4.0×10^{-8}
	$\gamma_e/2\pi$	28 024.9540(11)	MHz T ⁻¹	4.0×10^{-8}
Muon, μ^-				
muon mass	m_μ	$1.883\,531\,09(16) \times 10^{-28}$	kg	8.4×10^{-8}
in u, $m_\mu = A_r(\mu) u$ (muon relative atomic mass times u)		0.113 428 9168(34)	u	3.0×10^{-8}
energy equivalent in MeV	$m_\mu c^2$	$1.692\,833\,32(14) \times 10^{-11}$	J	8.4×10^{-8}
		105.658 3568(52)	MeV	4.9×10^{-8}
muon-electron mass ratio	m_μ/m_e	206.768 2657(63)		3.0×10^{-8}
muon-tau mass ratio	m_μ/m_τ	$5.945\,72(97) \times 10^{-2}$		1.6×10^{-4}
muon-proton mass ratio	m_μ/m_p	0.112 609 5173(34)		3.0×10^{-8}
muon-neutron mass ratio	m_μ/m_n	0.112 454 5079(34)		3.0×10^{-8}
muon molar mass $N_A m_\mu$	$M(\mu), M_\mu$	$0.113\,428\,9168(34) \times 10^{-3}$	kg mol ⁻¹	3.0×10^{-8}
muon Compton wavelength $h/m_\mu c$	$\lambda_{C,\mu}$	$11.734\,441\,97(35) \times 10^{-15}$	m	2.9×10^{-8}
$\lambda_{C,\mu}/2\pi$	$\tilde{\lambda}_{C,\mu}$	$1.867\,594\,444(55) \times 10^{-15}$	m	2.9×10^{-8}
muon magnetic moment	μ_μ	$-4.490\,448\,13(22) \times 10^{-26}$	$J T^{-1}$	4.9×10^{-8}
to Bohr magneton ratio	μ_μ/μ_B	$-4.841\,970\,85(15) \times 10^{-3}$		3.0×10^{-8}
to nuclear magneton ratio	μ_μ/μ_N	-8.890 597 70(27)		3.0×10^{-8}
muon magnetic moment anomaly				
$ \mu_\mu /(e\hbar/2m_\mu) - 1$	a_μ	$1.165\,916\,02(64) \times 10^{-3}$		5.5×10^{-7}
muon g-factor $-2(1 + a_\mu)$	g_μ	-2.002 331 8320(13)		6.4×10^{-10}
muon-proton				
magnetic moment ratio	μ_μ/μ_p	-3.183 345 39(10)		3.2×10^{-8}
Tau, τ^-				
tau mass ^d	m_τ	$3.167\,88(52) \times 10^{-27}$	kg	1.6×10^{-4}
in u, $m_\tau = A_r(\tau) u$ (tau)				

Fundamental Physical Constants — Atomic and Nuclear Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
relative atomic mass times u energy equivalent in MeV	$m_\tau c^2$	1.907 74(31) $2.847\ 15(46) \times 10^{-10}$ 1 777.05(29)	u J MeV	1.6×10^{-4} 1.6×10^{-4} 1.6×10^{-4}
tau-electron mass ratio	m_τ/m_e	3 477.60(57)		1.6×10^{-4}
tau-muon mass ratio	m_τ/m_μ	16.8188(27)		1.6×10^{-4}
tau-proton mass ratio	m_τ/m_p	1.893 96(31)		1.6×10^{-4}
tau-neutron mass ratio	m_τ/m_n	1.891 35(31)		1.6×10^{-4}
tau molar mass $N_A m_\tau$	$M(\tau), M_\tau$	$1.907\ 74(31) \times 10^{-3}$	kg mol^{-1}	1.6×10^{-4}
tau Compton wavelength $h/m_\tau c$ $\lambda_{C,\tau}/2\pi$	$\lambda_{C,\tau}$	0.697 70(11) $\times 10^{-15}$ 0.111 042(18) $\times 10^{-15}$	m m	1.6×10^{-4} 1.6×10^{-4}
proton mass in u, $m_p = A_r(p)$ u (proton relative atomic mass times u)	m_p	1.672 621 58(13) $\times 10^{-27}$	kg	7.9×10^{-8}
energy equivalent in MeV	$m_p c^2$	1.007 276 466 88(13) $1.503\ 277\ 31(12) \times 10^{-10}$ 938.271 998(38)	u J MeV	1.3×10^{-10} 7.9×10^{-8} 4.0×10^{-8}
proton-electron mass ratio	m_p/m_e	1 836.152 6675(39)		2.1×10^{-9}
proton-muon mass ratio	m_p/m_μ	8.880 244 08(27)		3.0×10^{-8}
proton-tau mass ratio	m_p/m_τ	0.527 994(86)		1.6×10^{-4}
proton-neutron mass ratio	m_p/m_n	0.998 623 478 55(58)		5.8×10^{-10}
proton charge to mass quotient	e/m_p	$9.578\ 834\ 08(38) \times 10^7$	C kg^{-1}	4.0×10^{-8}
proton molar mass $N_A m_p$	$M(p), M_p$	$1.007\ 276\ 466\ 88(13) \times 10^{-3}$	kg mol^{-1}	1.3×10^{-10}
proton Compton wavelength $h/m_p c$ $\lambda_{C,p}/2\pi$	$\lambda_{C,p}$	1.321 409 847(10) $\times 10^{-15}$ 0.210 308 9089(16) $\times 10^{-15}$	m m	7.6×10^{-9} 7.6×10^{-9}
proton magnetic moment to Bohr magneton ratio	μ_p	$1.410\ 606\ 633(58) \times 10^{-26}$	J T^{-1}	4.1×10^{-8}
to nuclear magneton ratio	μ_p/μ_N	$1.521\ 032\ 203(15) \times 10^{-3}$		1.0×10^{-8}
proton g-factor $2\mu_p/\mu_N$	g_p	2.792 847 337(29)		1.0×10^{-8}
proton-neutron magnetic moment ratio	μ_p/μ_n	-1.459 898 05(34)		2.4×10^{-7}
shielded proton magnetic moment (H ₂ O, sphere, 25 °C)	μ'_p	$1.410\ 570\ 399(59) \times 10^{-26}$	J T^{-1}	4.2×10^{-8}
to Bohr magneton ratio	μ'_p/μ_B	$1.520\ 993\ 132(16) \times 10^{-3}$		1.1×10^{-8}
to nuclear magneton ratio	μ'_p/μ_N	2.792 775 597(31)		1.1×10^{-8}
proton magnetic shielding correction $1 - \mu'_p/\mu_p$ (H ₂ O, sphere, 25 °C)	σ'_p	$25.687(15) \times 10^{-6}$		5.7×10^{-4}
proton gyromagnetic ratio $2\mu_p/\hbar$	γ_p	$2.675\ 222\ 12(11) \times 10^8$	$\text{s}^{-1} \text{T}^{-1}$	4.1×10^{-8}
shielded proton gyromagnetic ratio $2\mu'_p/\hbar$	γ'_p	$42.577\ 4825(18)$	MHz T^{-1}	4.1×10^{-8}
		$2.675\ 153\ 41(11) \times 10^8$	$\text{s}^{-1} \text{T}^{-1}$	4.2×10^{-8}

Fundamental Physical Constants — Atomic and Nuclear Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
(H ₂ O, sphere, 25 °C)				
	$\gamma_p'/2\pi$	42.576 3888(18)	MHz T ⁻¹	4.2×10^{-8}
		Neutron, n		
neutron mass in u, $m_n = A_r(n) u$ (neutron relative atomic mass times u)	m_n	1.674 927 16(13) × 10 ⁻²⁷	kg	7.9×10^{-8}
energy equivalent in MeV	$m_n c^2$	1.008 664 915 78(55) 1.505 349 46(12) × 10 ⁻¹⁰ 939.565 330(38)	u J MeV	5.4×10^{-10} 7.9×10^{-8} 4.0×10^{-8}
neutron-electron mass ratio	m_n/m_e	1 838.683 6550(40)		2.2×10^{-9}
neutron-muon mass ratio	m_n/m_μ	8.892 484 78(27)		3.0×10^{-8}
neutron-tau mass ratio	m_n/m_τ	0.528 722(86)		1.6×10^{-4}
neutron-proton mass ratio	m_n/m_p	1.001 378 418 87(58)		5.8×10^{-10}
neutron molar mass $N_A m_n$	$M(n), M_n$	1.008 664 915 78(55) × 10 ⁻³	kg mol ⁻¹	5.4×10^{-10}
neutron Compton wavelength $h/m_n c$ $\lambda_{C,n}/2\pi$	$\lambda_{C,n}$	1.319 590 898(10) × 10 ⁻¹⁵	m	7.6×10^{-9}
neutron magnetic moment to Bohr magneton ratio to nuclear magneton ratio	μ_n	0.210 019 4142(16) × 10 ⁻¹⁵ −0.966 236 40(23) × 10 ⁻²⁶ −1.041 875 63(25) × 10 ⁻³ −1.913 042 72(45)	m J T ⁻¹	7.6×10^{-9} 2.4×10^{-7} 2.4×10^{-7} 2.4×10^{-7}
neutron g-factor $2\mu_n/\mu_N$	g_n	−3.826 085 45(90)		2.4×10^{-7}
neutron-electron magnetic moment ratio	μ_n/μ_e	1.040 668 82(25) × 10 ^{−3}		2.4×10^{-7}
neutron-proton magnetic moment ratio	μ_n/μ_p	−0.684 979 34(16)		2.4×10^{-7}
neutron to shielded proton magnetic moment ratio (H ₂ O, sphere, 25 °C)	μ_n/μ'_p	−0.684 996 94(16)		2.4×10^{-7}
neutron gyromagnetic ratio $2 \mu_n /\hbar$	γ_n	1.832 471 88(44) × 10 ⁸	s ⁻¹ T ⁻¹	2.4×10^{-7}
	$\gamma_n/2\pi$	29.164 6958(70)	MHz T ⁻¹	2.4×10^{-7}
		Deuteron, d		
deuteron mass in u, $m_d = A_r(d) u$ (deuteron relative atomic mass times u)	m_d	3.343 583 09(26) × 10 ⁻²⁷	kg	7.9×10^{-8}
energy equivalent in MeV	$m_d c^2$	2.013 553 212 71(35) 3.005 062 62(24) × 10 ⁻¹⁰ 1.875.612 762(75)	u J MeV	1.7×10^{-10} 7.9×10^{-8} 4.0×10^{-8}
deuteron-electron mass ratio	m_d/m_e	3 670.482 9550(78)		2.1×10^{-9}
deuteron-proton mass ratio	m_d/m_p	1.999 007 500 83(41)		2.0×10^{-10}
deuteron molar mass $N_A m_d$	$M(d), M_d$	2.013 553 212 71(35) × 10 ⁻³	kg mol ⁻¹	1.7×10^{-10}
deuteron magnetic moment to Bohr magneton ratio to nuclear magneton ratio	μ_d	0.433 073 457(18) × 10 ⁻²⁶ 0.466 975 4556(50) × 10 ⁻³ 0.857 438 2284(94)	J T ⁻¹	4.2×10^{-8} 1.1×10^{-8} 1.1×10^{-8}

Fundamental Physical Constants — Atomic and Nuclear Constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
deuteron-electron magnetic moment ratio	μ_d/μ_e	$-4.664\,345\,537(50) \times 10^{-4}$		1.1×10^{-8}
deuteron-proton magnetic moment ratio	μ_d/μ_p	$0.307\,012\,2083(45)$		1.5×10^{-8}
deuteron-neutron magnetic moment ratio	μ_d/μ_n	$-0.448\,206\,52(11)$		2.4×10^{-7}
Helion, h				
helion mass ^c in u, $m_h = A_r(h)$ u (helion relative atomic mass times u)	m_h	$5.006\,411\,74(39) \times 10^{-27}$	kg	7.9×10^{-8}
energy equivalent in MeV	$m_h c^2$	$3.014\,932\,234\,69(86)$ $4.499\,538\,48(35) \times 10^{-10}$ $2\,808.391\,32(11)$	u J MeV	2.8×10^{-10} 7.9×10^{-8} 4.0×10^{-8}
helion-electron mass ratio	m_h/m_e	$5\,495.885\,238(12)$		2.1×10^{-9}
helion-proton mass ratio	m_h/m_p	$2.993\,152\,658\,50(93)$		3.1×10^{-10}
helion molar mass $N_A m_h$	$M(h), M_h$	$3.014\,932\,234\,69(86) \times 10^{-3}$	kg mol ⁻¹	2.8×10^{-10}
shielded helion magnetic moment (gas, sphere, 25 °C)	μ'_h	$-1.074\,552\,967(45) \times 10^{-26}$	J T ⁻¹	4.2×10^{-8}
to Bohr magneton ratio	μ'_h/μ_B	$-1.158\,671\,474(14) \times 10^{-3}$		1.2×10^{-8}
to nuclear magneton ratio	μ'_h/μ_N	$-2.127\,497\,718(25)$		1.2×10^{-8}
shielded helion to proton magnetic moment ratio (gas, sphere, 25 °C)	μ'_h/μ_p	$-0.761\,766\,563(12)$		1.5×10^{-8}
shielded helion to shielded proton magnetic moment ratio (gas/H ₂ O, spheres, 25 °C)	μ'_h/μ'_p	$-0.761\,786\,1313(33)$		4.3×10^{-9}
shielded helion gyromagnetic ratio $2 \mu'_h /\hbar$ (gas, sphere, 25 °C)	γ'_h	$2.037\,894\,764(85) \times 10^8$	s ⁻¹ T ⁻¹	4.2×10^{-8}
	$\gamma'_h/2\pi$	$32.434\,1025(14)$	MHz T ⁻¹	4.2×10^{-8}
Alpha particle, α				
alpha particle mass in u, $m_\alpha = A_r(\alpha)$ u (alpha particle relative atomic mass times u)	m_α	$6.644\,655\,98(52) \times 10^{-27}$	kg	7.9×10^{-8}
energy equivalent in MeV	$m_\alpha c^2$	$4.001\,506\,1747(10)$ $5.971\,918\,97(47) \times 10^{-10}$ $3\,727.379\,04(15)$	u J MeV	2.5×10^{-10} 7.9×10^{-8} 4.0×10^{-8}
alpha particle to electron mass ratio	m_α/m_e	$7\,294.299\,508(16)$		2.1×10^{-9}
alpha particle to proton mass ratio	m_α/m_p	$3.972\,599\,6846(11)$		2.8×10^{-10}
alpha particle molar mass $N_A m_\alpha$	$M(\alpha), M_\alpha$	$4.001\,506\,1747(10) \times 10^{-3}$	kg mol ⁻¹	2.5×10^{-10}

^a Value recommended by the Particle Data Group, Caso et al., Eur. Phys. J. C **3**(1-4), 1-794 (1998).

^b Based on the ratio of the masses of the W and Z bosons m_W/m_Z recommended by the Particle Data Group (Caso et al., 1998). The value for $\sin^2\theta_W$ they recommend, which is based on a particular variant of the modified minimal subtraction ($\overline{\text{MS}}$) scheme, is $\sin^2\hat{\theta}_W(M_Z) = 0.231\,24(24)$.

^c The helion, symbol h, is the nucleus of the ${}^3\text{He}$ atom.

^d This and all other values involving m_τ are based on the value of $m_\tau c^2$ in MeV recommended by the Particle Data Group, Caso et al., Eur. Phys. J. C **3**(1-4), 1-794 (1998), but with a standard uncertainty of 0.29 MeV rather than the quoted uncertainty of -0.26 MeV, $+0.29$ MeV.